

OPCW and the International Union of Pure and Applied Chemistry

**Reflecting on and Taking Forward a
Partnership with the Scientific Community**

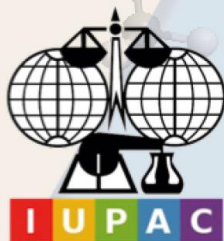
21 November

13:00 - 15:00

Antarctica Room

World Forum

Light lunch provided



INTERNATIONAL UNION
OF PURE AND APPLIED
CHEMISTRY



OPCW

Organisation for
the Prohibition of
Chemical Weapons

OPCW and the International Union of Pure and Applied Chemistry

**Reflecting on and Taking Forward a
Partnership with the Scientific Community**

Jonathan E. Forman, Ph.D.

**OPCW Science Policy Adviser and
Secretary to the Scientific Advisory Board**

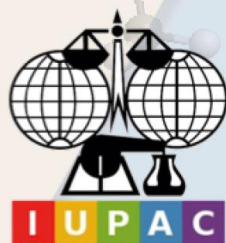
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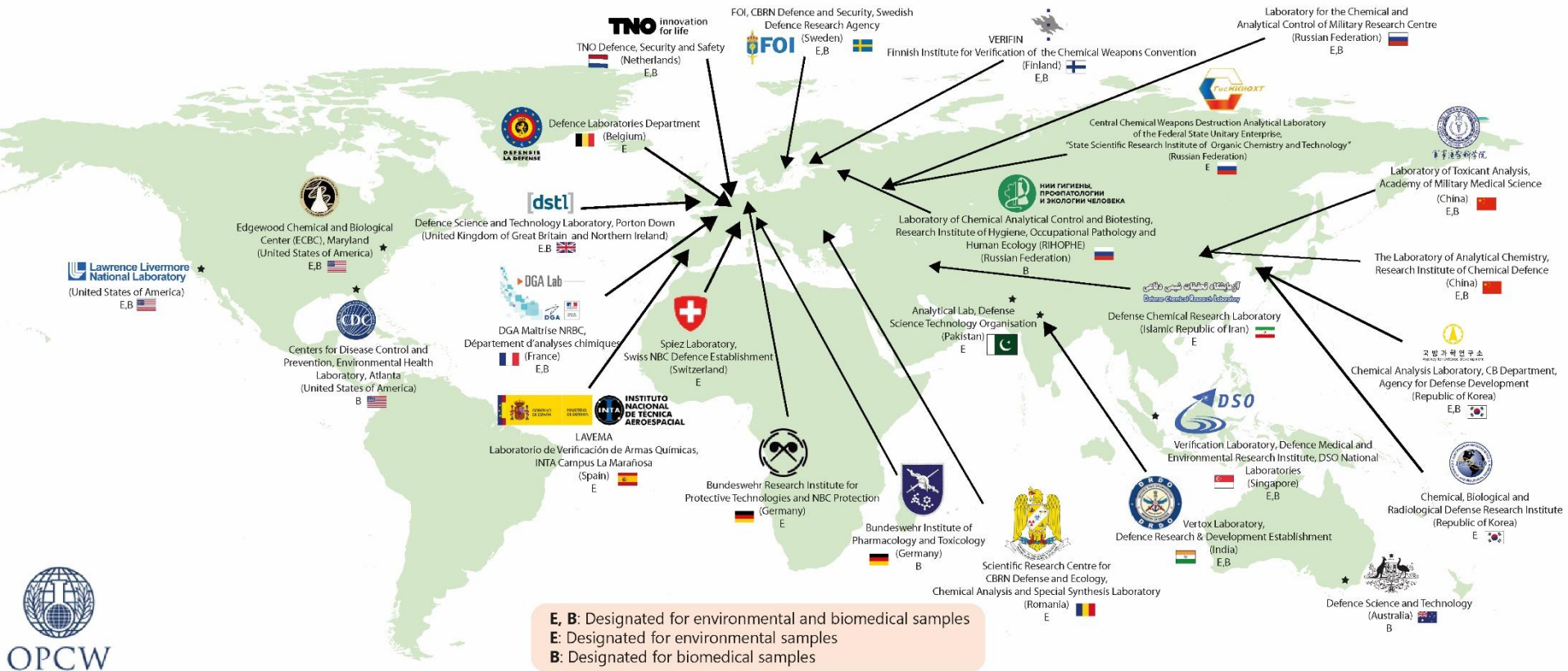
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the Prohibition of
Chemical Weapons

International Science Collaboration and the OPCW



OPCW

Technical Secretariat

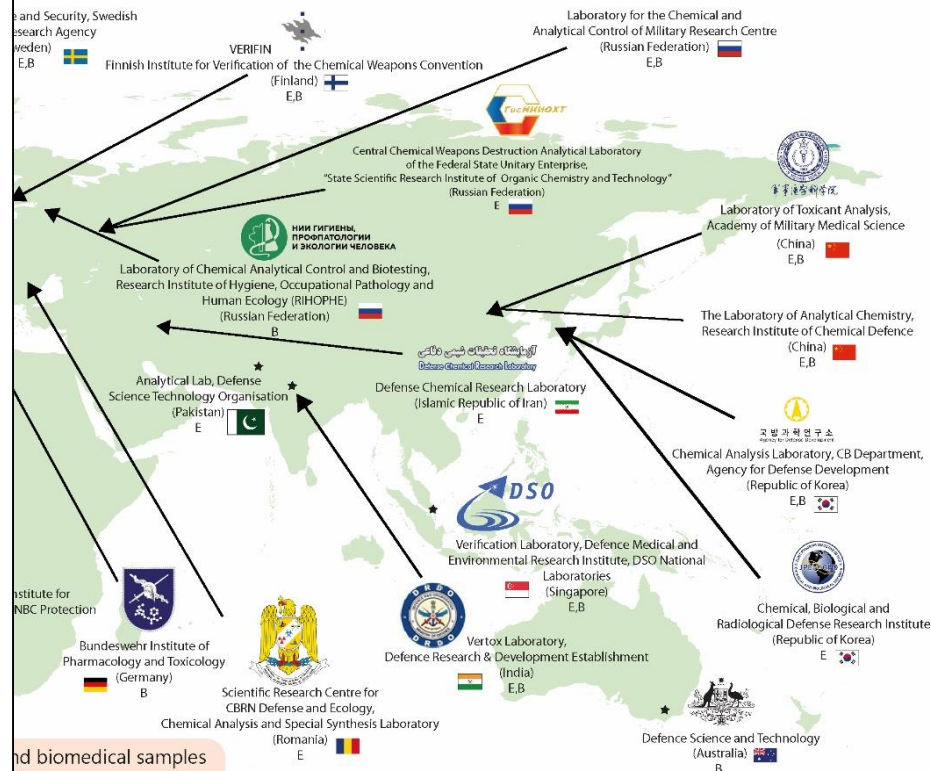
NOTE BY THE TECHNICAL SECRETARIAT

1. The Validation Group met on 25 and 26 September 2018 to discuss the evaluation of new analytical data for possible inclusion in the OPCW Central Analytical Database (OCAD) and to consider matters related to this database. Mr Hugh Gregg (United States of America) served as acting Chairperson of the meeting.
2. The evaluators for the analytical techniques evaluated new data and sent their written reports to the coordinators for each analytical technique. The names of the coordinators who were present at the meeting, along with the technique for which each was responsible, are listed below.

Mr Ferdinand Visser (South Africa)	Gas chromatography (retention index) (GC(RI))
Mr Vesa Häkkinen (acting coordinator) (Finland)	Mass spectrometry (MS)
Mr Armando Alcaraz (United States of America)	Infrared (IR) spectroscopy
Mr Damian Magiera (Germany)	Nuclear magnetic resonance (NMR) spectroscopy

3. The coordinators provided an evaluation summary of the data presented to the Validation Group for discussion at the meeting. The evaluators finalised the evaluation of the analytical data and confirmed that the approved data was technically valid.
4. The Validation Group accepted the resignation of Mr James Riches (United Kingdom of Great Britain and Northern Ireland) from his position as Chairperson of the Group and as member of the MS subgroup. It also accepted the resignations of Mr Alex Bennett (United Kingdom) from the MS subgroup and Mr Takeharu Wada (Japan) from the GC(RI) subgroup. The Group acknowledged the support of each of these contributors throughout the time of their membership.
5. The Validation Group welcomed Mr Arne Ficks (Germany) as a new member of the MS subgroup and Mr Harri August Heikkinen (Finland) as a new member of the NMR subgroup.

CS-2018-1379(E) distributed 25/10/2018



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International Science Collaboration and the OPCW



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Organisation for the Prohibition of Chemical Weapons

OPCW



Scientific Advisory Board 2018

(As of 11 July 2018)

Lawrence Livermore National Laboratory
(United States of America)
E.B.



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International Science Collaboration and the OPCW



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DE GRUYTER

Pure Appl. Chem. 2018; aop

Conference paper

Jonathan E. Forman*, Christopher M. Timperley, Siqing Sun and Darcy van Eerten

Chemistry and diplomacy

<https://doi.org/10.1515/pac-2018-0902>

Abstract: The Chemical Weapons Convention is a science-based international treaty for the disarmament and non-proliferation of chemical weapons. The Organisation for the Prohibition of Chemical Weapons (OPCW) serves as its implementing body. The treaty bans chemicals weapons, includes a verification mechanism to monitor compliance, and requires scientific and technical expertise for effective implementation. This necessitates a continuous engagement with scientific communities, whether informal or institutionalized (as demonstrated by the Designated Laboratories, Validation Group, and Scientific Advisory Board (SAB), of the OPCW), to ensure operation of the treaty keeps pace with scientific advances, and that enabling opportunities to meet challenges through scientific advances can be seized. The effective use of science for treaty implementation demands scientific literacy for decision making. Herein, the Convention, its scientific basis, need for scientific expertise, and mechanisms through which the OPCW engages scientists, are described. The function of the OPCW SAB, its review of science and technology to advise disarmament and non-proliferation policymakers, and its role in raising awareness of science within the world of international diplomacy, are reviewed.



MS subgroup
NMR subgroup

CS-2018-1379(E) distrib



Scientific Advisory Board 2018

(As of 11 July 2018)



OPCW

Implementing the Convention

The Chemical Weapons Convention

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graph TD; A[The Chemical Weapons Convention] --> B[Disarmament (Destruction and Verification)]; A --> C[Non-proliferation (Verification)]; A --> D[Assistance and Protection against CWs]; A --> E[International Cooperation];
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**Disarmament
(Destruction and
Verification)**

**Non-proliferation
(Verification)**

**Assistance and
Protection against CWs**

**International
Cooperation**

Implementing the Convention

STATEMENT

**International
Union of Pure and
Applied Chemistry**

OPCW CSP22

(verification)



**International
Cooperation**

**and
against CWs**

Implementing the Convention

OPCW-IUPAC MoU 2016

STATEMENT
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IUPAC and OPCW Collaboration Predates the MoU!

IUPAC organized workshops held in 2001, 2007 and 2012 formed the basis of the Reports of the Scientific Advisory Board to the First, Second and Third Review Conferences

Pure Appl. Chem., Vol. 74, No. 12, pp. 2323-2352, 2002.
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INTERNATIONAL UNION OF PURE AND APPLIED CHEMISTRY

IMPACT OF SCIENTIFIC DEVELOPMENTS ON THE CHEMICAL WEAPONS CONVENTION (IUPAC Technical Report)

Prepared for publication by
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EDWIN D. BECKER⁴
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Peace Studies, University of Bradford, Bradford, UK; ⁴National Institutes of Health, Bethesda, MD 20892, USA

International Advisory Board: IUPAC Workshop, Bergen, 2007
(UK): Wataru Ando (Japan); Josefa F. Buman (USA); Wil
Chen (China); Rita Cordeiro (Belgium); Claude Eon (France);
(UK): Folke Ingman (Sweden); Detlef Mating (Germany);
Pavson (UK); Boris Mysoskov (Russia); Norma S. Saldana (Norway);
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*Institutional affiliations of all authors included for identification and
corresponding author

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Pure Appl. Chem., Vol. 80, No. 1, pp. 175-200, 2008.
doi:10.1351/pac200880010175
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INTERNATIONAL UNION OF PURE AND APPLIED CHEMISTRY

IMPACT OF SCIENTIFIC DEVELOPMENTS ON THE CHEMICAL WEAPONS CONVENTION (IUPAC Technical Report)

Prepared for publication by
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International Advisory Board: IUPAC Workshop, Zagreb, 2007
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M. Isobe (Japan); R. Lubinski (France); J. Matusick (Canada);
(Russia); G. W. Parshall (USA); G. S. Pearson (UK); D. Skar
(Norway); L. F. Tette (Germany).

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Pure Appl. Chem., Vol. 85, No. 4, pp. 851-881, 2013.
http://dx.doi.org/10.1351/PAC-REP-12-11-18
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Impact of scientific developments on the Chemical Weapons Convention (IUPAC Technical Report)*

Katie Smallwood¹, Rolf Trapp², Robert Mathews³, Beat Schmidt⁴,
and Leiv K. Sydnès^{5,1}

¹Independent Consultant, Geneva, Switzerland; ²International Disarmament
Consultant, 74270 Chemsanaz, France; ³Defence Science and Technology
Organisation, Australia; ⁴Spiez Laboratory, 3700 Spiez, Switzerland; ⁵Department
of Chemistry, University of Bergen, 5007 Bergen, Norway

Abstract: This document represents the final report of discussions and conclusions arising
from the workshop on Developments in Science and Technology Relevant to the Chemical
Weapons Convention, held in Spiez, Switzerland in February 2012.

Keywords: Chemical Weapons Convention; CWC; implementation; science and technology;
Third Review Conference.

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 - 2.4 S&T advances specifically relevant to verification
 - 2.5 S&T advances specifically relevant to protection against CW
 - 2.6 The evolution of the international S&T environment
 - 2.7 Extending support for the CWC via outreach and education
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 - 3.7 Chemical safety and security: Possession, transfer, and acquisition
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 - 3.9 Chemical safety and security: Engaging the chemical sciences community

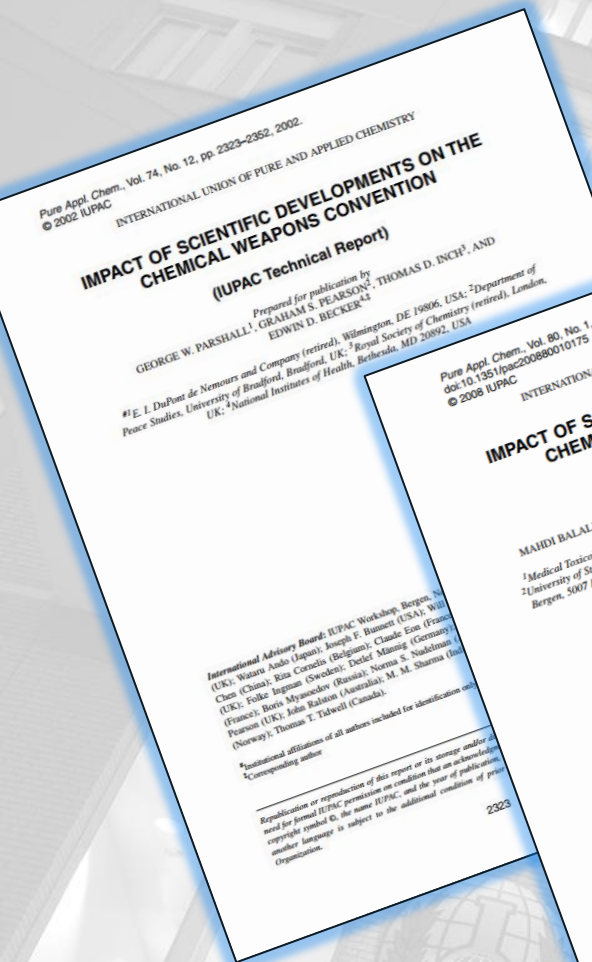
*Sponsoring body: IUPAC Executive Committee; see more details on p. 876.
Corresponding author: Email: leiv.sydnès@iuh.no

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IUPAC and OPCW Collaboration Predates the MoU!



03-05 | JULY - 2017
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Science for Peace

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Meeting Participation and Support

GREEN CHEMISTRY

17 to 21 August 2014

Durban, South Africa

2014

5th International IUPAC Conference on Green Chemistry

Co-hosted by:



5th IUPAC International Conference
on Green Chemistry ~ 2014

OPCW EVENTS

Wednesday 20 August

Plenary Lecture: Educating for a Sustainable and Secure Planet 8:30
Peter Mahaffy — Great Ilanga Room

Keynote: Green Chemistry Education in Africa 9:30
Engida Temechegn — Suites 1-3

OPCW Sustainability & Security Symposium

Security Dimensions of Sustainable Technology Development 9:30
Jonathan E. Forman — Great Ilanga Room

A Three-Legged Stool: IUPAC and OPCW Working Together to Promote Sustainability and Security 9:45
Peter Mahaffy & Alistair Hay — Great Ilanga Room

Sustainability, Security and Ethics 10:00
Liliana Mammino — Great Ilanga Room

OPCW Capacity Building Programmes for the Peaceful and Sustainable Applications of Chemistry 10:15
Sergey Zinoviev — Great Ilanga Room

Workshop: Multiple Uses of Chemicals 13:30
Peter Mahaffy & Alistair Hay — Great Ilanga Room

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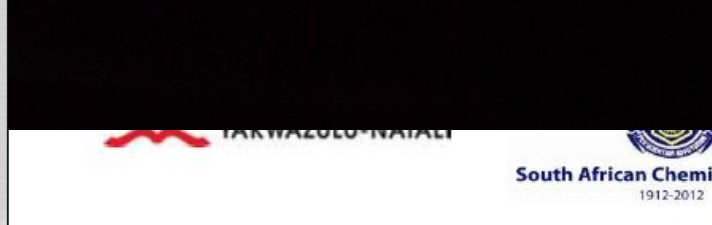
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International Conference Chemistry ~ 2014 EVENTS

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Sustainable Technology	9:30



IUPAC 2017
São Paulo, Brazil

46th World Chemistry Congress
40^a Reunião Anual da Sociedade Brasileira de Química
July 9 to 14, 2017 - São Paulo - Brazil

IUPAC 49th General Assembly
July 7 to 13, 2017 - São Paulo - Brazil

Sustainability & Diversity through Chemistry

Peter Manary & Alistair Hay — Great Hall A Room

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S/1649/2018
10 July 2018
ENGLISH only

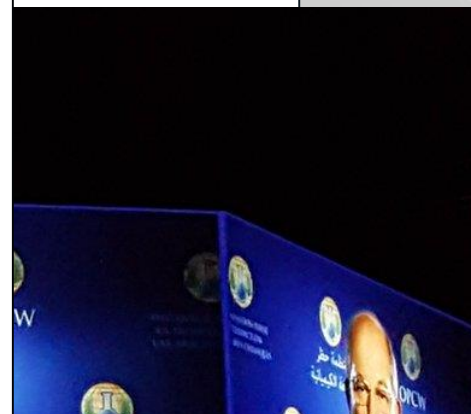
NOTE BY THE TECHNICAL SECRETARIAT

**INVITATION TO APPLY FOR A FORUM ON THE PEACEFUL USES OF CHEMISTRY:
POTENTIAL CONTRIBUTION OF THE OPCW TO THE ACHIEVEMENT OF THE
UNITED NATIONS SUSTAINABLE DEVELOPMENT GOALS
THE HAGUE, THE NETHERLANDS
26 OCTOBER 2018**

Purpose of the forum

1. The Technical Secretariat of the Organisation for the Prohibition of Chemical Weapons (hereinafter "the Secretariat") wishes to inform Member States that it is organising a forum on the potential contribution of the OPCW to the achievement of the United Nations Sustainable Development Goals. Special focus will be on the potential contribution of international cooperation in the field of chemicals and Sustainable Development Goals. The forum will be held at the OPCW Headquarters in The Hague, The Netherlands, on 26 October 2018.
2. The forum is open to government officials, scientists, academics and industry representatives working in the field of chemicals and Sustainable Development Goals.
3. The overall objective of the forum is to promote international cooperation and the United Nations Sustainable Development Goals through the following strategic objectives:
 - (a) presenting and discussing the role of chemicals in the achievement of the Sustainable Development Goals;
 - (b) discussing the synergies between the Sustainable Development Goals and the chemical industry;
 - (c) identifying potential partnerships and stakeholders within the chemical industry relating to topics that fall within the scope of the Sustainable Development Goals;
 - (d) exploring opportunities for the chemical industry to contribute to the achievement of the Sustainable Development Goals.

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International Conference
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Symposium

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IUPAC
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PURE AND APPLIED CHEMISTRY



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de Química

FOR THE
PROHIBITION OF CHEMICAL WEAPONS



OPCW

Committees and Working Groups

WHO WE ARE

WHAT WE DO

EVENTS

PROJECTS

NEWS

WHO WE ARE

MEMBER DIRECTORY

OUR LEADERSHIP

SECRETARIAT

OUR HISTORY

STRATEGIC PLAN

ORGANIZATIONAL CHART

DIVISIONS

COMMITTEES

GOVERNANCE

NATIONS AFFILIATED WITH IUPAC

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IUPAC Standing Committees are established by the [Council](#) to serve as Advisory Bodies for the President and the [Executive Committee](#). The current Committees are as follows:

[Bureau](#)

[Chemistry International Editorial Board](#)

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[Committee on Chemistry and Industry](#)

[Committee on Chemistry Education](#)

[Committee on Publications and Cheminformatics Data Standards](#)

[Evaluation Committee](#)

[Executive Committee](#)

[Finance Committee](#)

[Interdivisional Committee on Green Chemistry for Sustainable Development \(ICGCSD\)](#)

[Interdivisional Committee on Terminology, Nomenclature and Symbols](#)

[Project Committee](#)

[Pure and Applied Chemistry Editorial Advisory Board](#)



OPCW

Committees and Working Groups




EDUCATION AND ENGAGEMENT:
Promoting a Culture of Responsible Chemistry


**FINAL REPORT OF THE SCIENTIFIC ADVISORY
BOARD'S TEMPORARY WORKING GROUP**

NOVEMBER 2014



**ORGANISATION FOR THE PROHIBITION
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- [Chemistry Education](#)
- [Publications and Cheminformatics Data Standards Committee](#)
- [Committee on Green Chemistry for Sustainable Development \(ICGCSD\)](#)
- [Committee on Terminology, Nomenclature and Symbols](#)
- [Global Chemistry Editorial Advisory Board](#)



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Committees and Working Groups



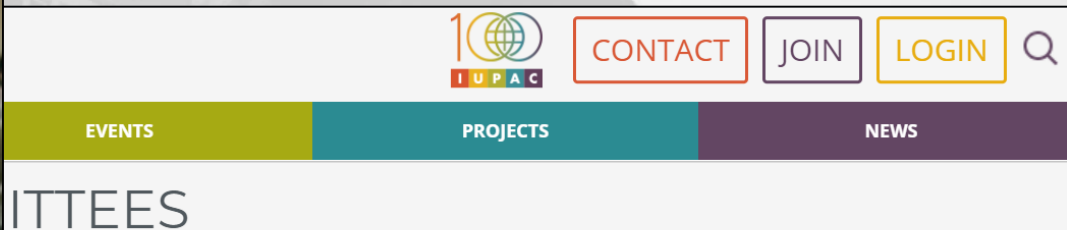
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Promoting OPCW in Scientific Literature




CCCE Members ConfChem Archive Newsletter Archive Archives ConfChem and Newsletter Discussion Protocols

DivCHED CCCE: Committee on Computers in Chemical Education

[Home](#)

2016 Spring ConfChem: Science, Disarmament, and Diplomacy in Chemical Education: The Example of the Organisation for the Prohibition of Chemical Weapons

05/02/16 to 07/01/16



Science plays a critical role in international disarmament policy and diplomacy; informing the negotiations of international agreements, and underpinning their articles and implementation. Of relevance to the science of chemistry, the disarmament of chemical weapons provides an opportunity to introduce students to the nexus of science and international diplomacy.

An international disarmament treaty banning chemical weapons, the Chemical Weapons Convention (CWC), has been in force since 1997. The treaty has 192 States Parties (the governments that have agreed to uphold the norms and obligations required by the treaty) and is implemented by the Organisation for the Prohibition of Chemical Weapons (OPCW); an international organization located in The Hague in the Netherlands and the recipient of the 2013 Nobel Peace Prize for its efforts in chemical disarmament.

The OPCW has existed since 1997 and has overseen the destruction of more than 90% of the world's declared military stockpiles of chemical weapons. The CWC is the most widely subscribed disarmament treaty in history. But the organization has not been widely recognized outside disarmament-focused diplomatic circles. This is in spite of the fact that the science of chemistry is an integral part of the CWC and the work of the OPCW, requiring that the organization interact with the greater chemistry community. In order to achieve its goals in the future, the OPCW will need to up its game in this area, reaching out to new stakeholders and strengthening its ties with its existing partners. To this end, the OPCW has been placing an increasing priority on education and engagement to raise awareness of its work and the contributions both from and to science in chemical disarmament.

This ConfChem Online Conference is designed to highlight the work of the OPCW and the important contribution of scientists and educators to achieving its goals, the science that underpins the Chemical Weapons Convention, and how scientific and technological advances will help to better implement the Convention in the future.

Papers and Discussion Schedule

May 2-6: **Education, outreach and the OPCW: growing partnerships for a global ban**

May 9-13 **Education and Engagement: Key Elements to Achieve a World Free of Chemical Weapons**

May 16-20 **Mainstreaming Multiple Uses of Chemicals in Chemistry Teacher Education Programs of Africa**

May 23-27 **The project Irresistible: Introducing Responsible Research and Innovation into the Secondary School Classroom**

May 30-June 3: **Citizen Science and International Collaboration through Environmental Monitoring with Simple Chemical Sensors**

June 6-10: **Painful chemistry! From barbeque smoke to riot control**

June 13-17: **Sampling and Analysis of Organophosphorus Nerve Agents: Analytical Chemistry in International Chemical Disarmament**

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Papers and Discussion Schedule

May 2-6: [Education, outreach and the](#)
May 9-13: [Education and Engagement:](#)
May 16-20: [Mainstreaming Multiple Uses](#)
May 23-27: [The project Irresistible: Intro](#)
May 30-June 3: [Citizen Science and Inter](#)
June 6-10: [Painful chemistry! From barl](#)
June 13-17: [Sampling and Analysis of Or](#)



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**INTERNATIONAL WORKSHOP
ON INNOVATIVE TECHNOLOGIES
FOR CHEMICAL SECURITY**

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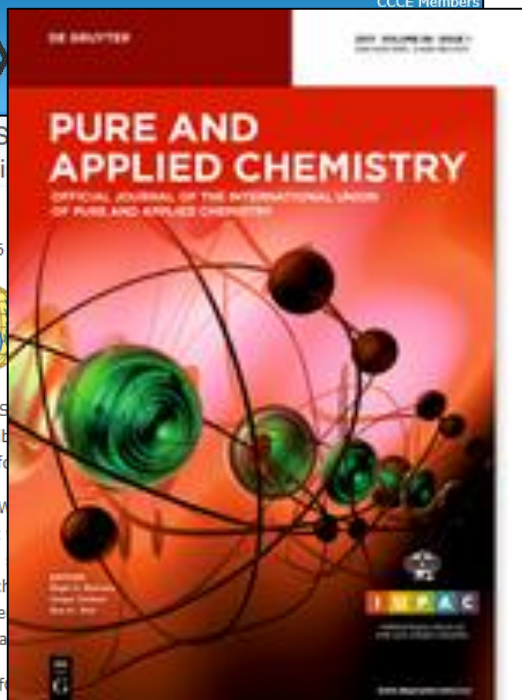
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Papers and Discussion Schedule

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- May 9-13: **Education and Engagement:**
- May 16-20: **Mainstreaming Multiple Uses**
- May 23-27: **The project Irresistible: Intro**
- May 30-June 3: **Citizen Science and Inter**
- June 6-10: **Painful chemistry! From barl**
- June 13-17: **Sampling and Analysis of Or**



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Pure Appl. Chem. 2017; aop

Conference paper

Robert E. Belford* and Jonathan E. Forman

Science, disarmament and diplomacy in chemical education: the example of the organisation for the prohibition of chemical weapons – The Spring 2016 Confchem

DOI 10.1515/pac-2016-1115

Abstract: In May and June of 2016 the Organisation for the Prohibition of Chemical Weapons (OPCW), IUPAC and the ACS CHED Committee on Computers in Chemical Education (CCCE) collaboratively ran an online conference hosted with the online ConfChem conference system on "Science, Disarmament and Diplomacy in Chemical Education".

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Pure Appl. Chem. 2018; aop

Preface

Mark C. Cesa, Vitor Francisco Ferrelira, Jonathan E. Forman*, Cheng Tang, Christopher M. Timperley, Camly Tran and Bernard West

OPCW-IUPAC Workshop on Innovative Technologies for Chemical Security

<https://doi.org/10.1515/pac-2018-0701>

Abstract: The Organisation for the Prohibition of Chemical Weapons (OPCW), the International Union of Pure and Applied Chemistry (IUPAC), The National Academies of Science, Engineering and Medicine of the USA, the Brazilian Academy of Sciences, and the Brazilian Chemical Society held a workshop. "Innovative Technologies for Chemical Security", in Rio de Janeiro, Brazil, from 3 to 5 July 2017. This event was part of a four workshop series held to inform the report of the OPCW Scientific Advisory Board on developments in science and technology to the Fourth Review Conference of the Chemical Weapons Convention, which will be held in November 2018. The workshop explored the potential of new technologies to enhance capabilities for the implementation of the Chemical Weapons Convention. There is a continuing need for recognition that emerging scientific developments can have beneficial applications with respect to implementation of the Convention, particularly in prevention of re-emergence of chemical weapons. The objectives of this workshop were to present, discuss and critically evaluate the emergence and practical applications of new and existing technologies – as tools for detecting biochemical change in complex environments – and the applications of these technologies in support of chemical disarmament and chemical security. This issue of *Pure and Applied Chemistry* presents a series of papers that originate from topics discussed in the workshop. This preface describes the scientific review process for the Chemical Weapons Convention and how it was supported by the Rio de Janeiro workshop, as well as introducing the papers in the collection and their corresponding authors.



by



IN OF
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Ethical Codes and Guidelines

Pure Appl. Chem., Vol. 78, No. 11, pp. 2169–2192, 2006.
doi:10.1351/pac200678112169
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INTERNATIONAL UNION OF PURE AND APPLIED CHEMISTRY

COMMITTEE ON CHEMISTRY EDUCATION*

EDUCATION, OUTREACH, AND CODES OF CONDUCT TO FURTHER THE NORMS AND OBLIGATIONS OF THE CHEMICAL WEAPONS CONVENTION

(IUPAC Technical Report)

Prepared for publication by
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The 2005 Oxford Workshop leading to this technical report was organized by OPCW's R. Trapp and members of IUPAC Project 2004-048-1-020 (a joint OPCW/IUPAC Project on Education and Outreach regarding Chemical Weapons) P. W. Atkins (UK); E. D. Becker (USA); L. K. Sydnos (Norway); and N. P. Tarasova (Russia).

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OPCW

Ethical Codes and Guidelines

Pure Appl. Chem., Vol. 78, No. 11, pp. 2169–2192, 2006.
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ENSURING AN ETHOS OF SCIENCE FOR PEACE THE HAGUE ETHICAL GUIDELINES

IUPAC contributed
and endorsed



OPCW

Taking a Partnership to the Next Level

RECENT POSTS

On the revision of the International System of Units
Launch of the 2019 International Year of the Periodic
Table of Chemical Elements
On the discovery of new elements
eTOC Alert 'Chemistry International' – Oct-Dec 2018
OPCW to Further Enhance Contributions to United
Nations' Sustainable Development Goals

CATEGORIES

UPCOMING DEADLINES

AWARDS & PRIZES

GRANTS

UTHINGS

UPCOMING EVENTS

ANNOUNCEMENTS

CALL FOR INPUT

RECENT RELEASES

TAGS

aromatic compounds art & science big data bioorganic chemistry
boron chemistry chemical identifier chemometrics chitin CITAC
computational biochemistry CPCDS data data standards
emerging technology genomics green
chemistry herbal medicine heterocyclic chemistry

IUPAC AND THE ORGANISATION FOR THE PROHIBITION OF CHEMICAL WEAPONS TAKE PARTNERSHIP TO NEW LEVEL

1 December 2016



The Director-General of the Organisation for the Prohibition of Chemical Weapons (OPCW), Ambassador Ahmet Üzümcü, and IUPAC President Professor Natalia Tarasova, signed a Memorandum of Understanding (MOU) today pledging to enhance cooperation to keep abreast of developments in chemistry, responsibility and ethics in science, and education and outreach.

Ambassador Üzümcü remarked, "Promoting responsible science is a crucial endeavour to advance the goals of the Chemical Weapons Convention. Without scientists, there is no disarmament. IUPAC's unwavering commitment to a world permanently free of chemical weapons demonstrates the strength of this norm and the conviction of chemistry researchers and practitioners globally to protect it".

Professor Tarasova expressed, "Through the cooperation between our organisations, we look to help humanity achieve Sustainable Development Goals in a world free of chemical weapons and in a world in which achievements in chemical science and technology are used only for the benefit of humankind and the environment".

The MOU opens a new chapter and underscores the long-standing and productive relationship between the OPCW and IUPAC.

Background

As the implementing body for the Chemical Weapons Convention, the OPCW oversees the global endeavour to permanently eliminate chemical weapons. Since the Convention's entry into force in 1997 – with its 192 States Parties – it is the most successful disarmament treaty eliminating an entire class of weapons of mass destruction.



OPCW Director-General Ahmet Üzümcü (right) and IUPAC President Natalia Tarasova signed a MOU pledging to enhance cooperation to keep abreast of developments in chemistry, responsibility and ethics in science, and education and outreach. (1 Dec 2016)

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IUPAC-OPCW Collaboration

Prof Richard Hartshorn – IUPAC Secretary General
and members of the IUPAC Delegation



IUPAC is an indispensable resource for chemistry.



I U P A C MISSION

The International Union of Pure and Applied Chemistry is the global organization that provides objective scientific expertise and develops the essential tools for the application and communication of chemical knowledge for the benefit of humankind and the world.



I U P A C MISSION

The International Union of Pure and Applied Chemistry accomplishes its mission by

- Fostering sustainable development
- Providing a common language for chemistry
- Advocating the free exchange of scientific information



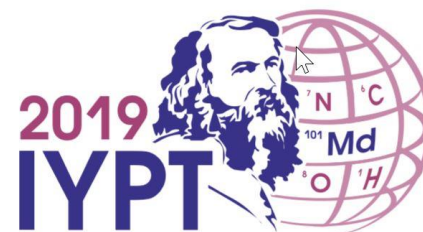
Two Global Celebrations in 2019



100 YEARS of CREATING A
COMMON LANGUAGE FOR
CHEMISTRY



United Nations
Educational, Scientific and
Cultural Organization



International Year
of the Periodic Table
of Chemical Elements

IUPAC in it's First Century

Development of the Language of Chemistry

- Nomenclature, Symbols, Terminology

Standardisation of Chemistry Methods

- Data Presentation
- Study of Analytical Methods

Critical Evaluation of Physico-Chemical Data

- Atomic Weights
- Thermodynamic Data
- Kinetic Data



More IUPAC Activities

- Data Exchange Standards for Computers and Instruments
- Endorsement of International Conferences
 - Biennial IUPAC Congress
 - More than 30 Specialised Symposia each Year
- Chemistry Education
- Industrial Safety and Environmental Programs
- ChemRAWN Conferences addressing Chemistry and World Needs
- ICGCSD Conferences and Summer Schools on Green Chemistry



IUPAC's Color Books



Chemical Terminology	Gold Book
Quantities Units and Symbols in Physical Chemistry	Green Book
Nomenclature in Organic Chemistry	Blue Book
Compendium of Polymer Terminology and Nomenclature	Purple Book
Analytical Terminology	Orange Book
Biochemical Terminology	White Book
Nomenclature in Inorganic Chemistry	Red Book
Compendium of Terminology and Nomenclature of Properties in Clinical Laboratory Science	Silver Book

What Will Our Second Century Look Like?

Big Data and the Rise of Cheminformatics

GO FAIR and the Chemistry Implementation Network (ChIN)

Tools to Support This:

- International Chemical Identifier (InChI and InChIKey)
- Reaction InChI and InChI for Mixtures
- Spectroscopic Data Standards (e.g. JCAMP-DX)

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IUPAC-OPCW Collaboration

Prof Ron Weir – President IUPAC Division I
Physical and Biophysical Chemistry

Provides the world standard for clear definitions and meaning of terminology, nomenclature, symbols and units

Encourages compilation and documentation of critically evaluated physical, biophysical, biochemical chemical data including those related to chemical weapons, their detection and disposal

Promote future oriented activities in physical, biophysical, biochemistry chemistry important for sustainable development

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IUPAC-OPCW Collaboration

Prof Jan Reedijk – IUPAC Division II
Inorganic Chemistry

IUPAC Division II – Inorganic Chemistry

The Division's **major areas** of focus are (1) *isotopic abundances and atomic weights*; (2) *molecular inorganic chemistry*; and (3) *solid-state inorganic and materials chemistry*.

Commission on Isotopic Abundances and Atomic Weights (CIAAW).
3 Subcommittees: Isotopic Abundance Measurements; Stable Isotope Reference Material Assessment; Natural Assessment of Fundamental Understanding of Isotopes.

Interactive online version of the Periodic Table has been made and updated, as well as the Periodic Table of Isotopes.

Interdivisional Subcommittee on Materials Chemistry

The Inorganic Chemistry Division is also the body within IUPAC that works with IUPAP on the **verification of claims** for and the naming of **new chemical elements**.

4 New Chemical Elements recognized and named in 2017 (Nh, Mc, Ts, Og)

Even though such new elements have little current applicability, the approval of the discoveries and the naming of new elements is **a highly visible activity for IUPAC** that attracts significant public attention to the IUPAC Periodic Table of the Elements and in general for IUPAC.

In 2019: Celebration of the UN choice:
International Year of the Periodic Table. See:

www.iypt2019.org

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IUPAC-OPCW Collaboration

Prof Nikolay Nifantiev – IUPAC Division III
Organic and Biomolecular Chemistry

IUPAC Division III - Organic and Biomolecular Chemistry



Division III could contribute to IUPAC-OPCW collaboration providing advice in :

- Defining criteria for selecting new toxicants and their cataloguing
- Selecting organic compounds for cataloguing as CW
- Selecting CW's metabolites (including bio-adducts) and destruction products for cataloguing
- Selecting potential antidotes that could be used
- Development of CWs destroying protocols and ecological monitoring

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IUPAC-OPCW Collaboration

Marloes Peeters – IUPAC Division IV

Polymer Division

Polymer Division (IV)

Polymer: large molecule, composed of many repeating units

Polymer Terminology

- Defining standards : adding IUPAC boxes to Wiki
 - >75 boxes added, plus concepts defined
- **New project OPCW: terminology / overview sensors**

Modeling Polymerization Kinetics and Processes

- Critical paper: How fast is polymerization?
- >500 citations

Polymer Education

- Free resources : website (different languages)
- Exchanging best practice

Subcommittee on Properties of Commercial Polymers

- Example projects: biodegradable materials
- Defining standard (ISO) procedures

IUPAC definition

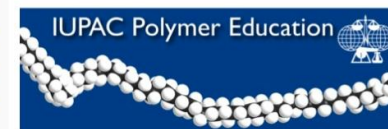
A molecule of high relative molecular mass, the structure of which essentially comprises the multiple repetition of units derived, actually or conceptually, from molecules of low relative molecular mass.^[2]

IUPAC Polymer Education Website

Changing the world, one polymer at a time...



- Home
- What are polymers?
- Polymer Naming
- Polymer Videos
- Polymer Glossary
- Int Yr of Chem – IYC 2011
- Int. Funding Call
- Polymer Ed Symposium
- Educational Web Sites



This website is intended to serve as a means for sharing educational tools dedicated to polymer chemistry. We offer free downloads, connections to other polymer education web sites and materials for teaching about polymers.

A Brief Guide to Polymer Nomenclature (Polymer Naming)

This two-page document shows how to name a polymer using IUPAC recommendations. It is a basic guide.....

PVC



POLYSTYRENE



PET



POLYURETHANE



PERSPEX



POLYCARBONATES

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IUPAC-OPCW Collaboration

Prof Hemda Garelick – IUPAC Division VI
Chemistry and the Environment

IUPAC Division VI - Chemistry and the Environment

We share the vision of OPCW especially on Chemistry for Peace and the sound management of chemicals.

The Division provides authoritative reviews and guidance on the fate, behaviour and risks of chemical compounds in food and the environment. It has an extensive network of experts around the globe.

Collaboration with OPCW should encompass :

- Risk assessment of chemicals
- The environmental footprints of chemical and radioactive materials/weapons as well as their transformation products – during manufacture and if used.
- Safe disposal of restricted chemicals



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IUPAC-OPCW Collaboration

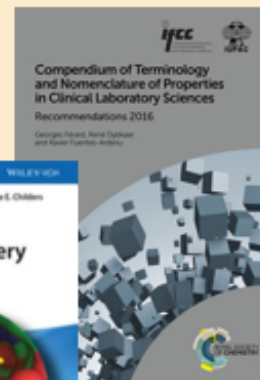
Michael Schwenk – IUPAC Division VII
Chemistry and Human Health

IUPAC Division VII

Chemistry and Human Health

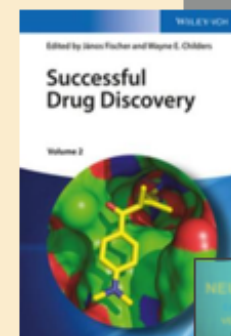
Nomenclature for Properties and Units:

Development of a universal terminology in Clinical Chemistry



Drug Discovery and Development:

International activities to promote the design of novel therapeutic drugs.



Toxicology and Risk Assessment:

Compilation of glossaries as worldwide basis for chemical risk assessment.



Possible Areas for Cooperation with OPCW

Global protection from chemical hazards and ban of chemical weapons.

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IUPAC-OPCW Collaboration

Prof Ed Constable – IUPAC Division VIII

Chemical Nomenclature and Structure Representation

IUPAC Division VIII - Nomenclature

Development of unique and interoperable nomenclature for chemical materials

- Unique and unambiguous identification

Interface with databanks for properties through:

- Name
- Computer readable unique descriptor (InChITM or SMILES)
- Structural formula

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IUPAC-OPCW Collaboration

Prof Jan Apotheker – IUPAC CCE
Committee for Chemistry Education

IUPAC Committee on Chemistry Education

Organization of educational researchers and lecturers

- To develop relationships for working collaboratively with groups both inside and outside of IUPAC presentation of the IUPAC activities in the implementation of UN SDGs and of SAICM
- To continue supporting initiatives that raise awareness, social responsibility, and understanding the nature of science as well as of environmental and ethical issues that are related to chemistry;
- To initiate programs on promoting chemistry education and public understanding of chemistry for developing countries;

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IUPAC-OPCW Collaboration

Prof Pietro Tundo – IUPAC ICGCSD

Interdivisional Committee on Green Chemistry for
Sustainable Development

Interdivisional Committee on Green Chemistry for Sustainable Development, ICGCSD

The Interdivisional Committee initiates, promotes, and coordinates the work of the Union in the area of green and sustainable chemistry.

ICGCSD is responsible for the promulgation of the work of the Union in green and sustainable chemistry through interaction with other relevant international chemical and non-chemical organizations with a common interest.

IUPAC/Green Chemistry and OPCW

Peaceful utilization of Chemistry and the 17UNSDGs

1) Green reagents & Substitution of harmful chemicals

Practical Outcome: reduced needs of inspections

2) Chemical Safety through the Management of Chemicals

Responsible Care – Responsible Utilisation of Chemistry

Metrics, Standardization, LCA, Risk assessment

Regulatory Strategies, Enabling Policies

Practical Outcomes: SELF CONTROL on chemical manufacture

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IUPAC-OPCW Collaboration

Prof Anna Makarova – IUPAC COCI
Committee for Chemistry and Industry

IUPAC Committee on Chemistry and Industry

Organization of scientists and a chemical industry collaboration

- development of a **Safety Training Programs** and **training for the implementation Responsible Care Program** (area of interest are developing countries: CEE, Central Asia, Africa)
- presentation of the IUPAC activities in the implementation of UN SDGs and of SAICM
- development together with ICGCSD of systems of criteria, standards, labeling and certification for assessing the compliance of chemical enterprises with the principles of Green Chemistry.





OPCW

Organisation for the Prohibition of Chemical Weapons



United Nations
Educational, Scientific and
Cultural Organization



International Year
of the Periodic Table
of Chemical Elements

International Year of the Periodic Table of Chemical Elements (IYPT) 2019

Jo-Anne Rasmussen PhD, Inspector, Inspectorate Division
Technical Secretariat
21 November 2018

History

- The Russian chemist Dmitri Mendeleev was the first scientist to make a periodic table similar to the one used today.
- In 1869, the table was published in an obscure Russian journal and then republished in a German journal, *Zeitschrift für Chemie*.
- Built upon earlier discoveries by scientists such as Antoine-Laurent de Lavoisier and John Newlands, but who is nevertheless generally given sole credit for its development.
- Henry Moseley in 1914 created the new order that was in agreement with the chemical properties of these elements.

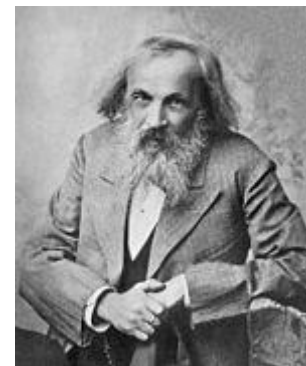


TABLEAU DES SUBSTANCES SIMPLES.

	Noms nouveaux.	Noms anciens correspondans.
	Lumière.....	Lumière. Chaleur. Principe de la chaleur.
	Calorique.....	Fluide igné. Feu. Matière du feu &c de la chaleur.
Substances simples qui appartiennent aux trois règnes & qu'on peut regarder comme les élémens des corps.	Oxygène.....	Air déphlogistiqué. Air empiréal. Air vital. Bâse de l'air vital.
	Azote.....	Gaz phlogistiqué. Mofette. Bâse de la mofette.
	Hydrogène.....	Gaz inflammable. Bâse du gaz inflammable.
Substances simples non métalliques oxidables & acidifiables.	Soufre.....	Soufre.
	Phosphore.....	Phosphore.
	Carbone.....	Charbon pur.
	Radical muriatique.	Inconnu.
	Radical fluorique.	Inconnu.
	Radical boracique.	Inconnu.
	Antimoine.....	Antimoine.
	Argent.....	Argent.
	Arsenic.....	Arsenic.
	Bismuth.....	Bismuth.
	Cobalt.....	Cobalt.
	Cuivre.....	Cuivre.
	Etain.....	Etain.
	Fer.....	Fer.
	Manganèse.....	Manganèse.
	Mercure.....	Mercure.
	Molybdène.....	Molybdène.
	Nickel.....	Nickel.
	Or.....	Or.
	Platine.....	Platine.
	Plomb.....	Plomb.
	Tungstène.....	Tungstène.
	Zinc.....	Zinc.
Substances simples métalliques oxidables & acidifiables.	Chaux.....	Terre calcaire, chaux.
	Magnésie.....	Magnésie, bâse du sel d'Épſom.
	Baryte.....	Barote, terre pesante.
	Alumine.....	Argile, terre de l'alun, bâse de l'alun.
Substances simples salifiables terreneſes.	Silice.....	Terre ſiliceuſe, terre vitrifiable.

VIS TELLURIQUE

CLASSEMENT NATUREL DES CORPS SIMPLES OU RADICAUX

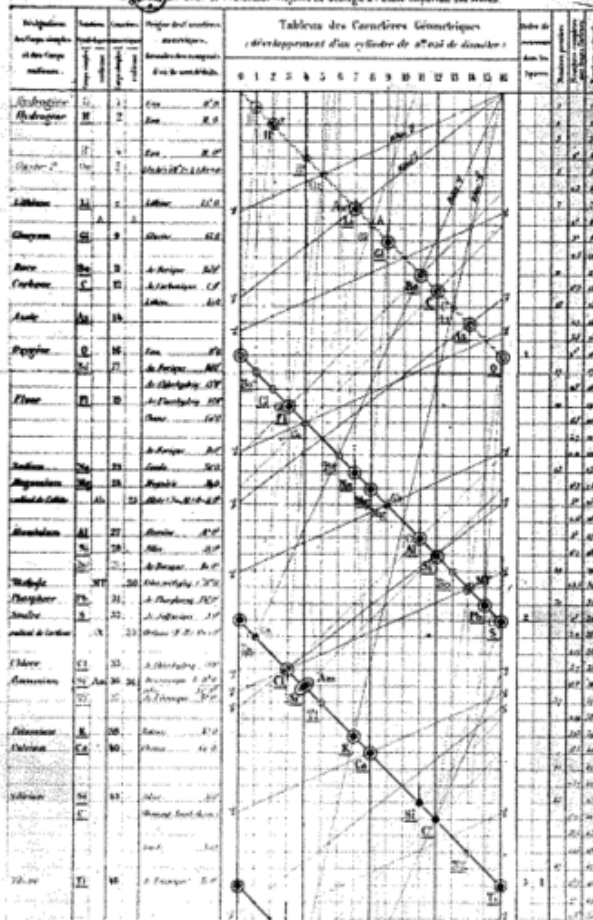
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SYSTEME DE CLASSIFICATION HELICOIDAL ET NUMERIQUE

A. E. BÉCUYER et CHANCOURTOIS

de la Promotion de 1834 à l'École Polytechnique

Ingénieur en Chef et Professeur-Adjoint de Géologie à l'École Supérieure des Mines

[illegible]

Исходъ и възвращеніе

$\begin{array}{l} \text{A=1} \\ \text{B=2} \\ \text{C=3} \\ \text{D=4} \\ \text{E=5} \\ \text{F=6} \\ \text{G=7} \\ \text{H=8} \\ \text{I=9} \\ \text{J=10} \\ \text{K=11} \\ \text{L=12} \\ \text{M=13} \\ \text{N=14} \\ \text{O=15} \\ \text{P=16} \\ \text{Q=17} \\ \text{R=18} \\ \text{S=19} \\ \text{T=20} \\ \text{U=21} \\ \text{V=22} \\ \text{W=23} \\ \text{X=24} \\ \text{Y=25} \\ \text{Z=26} \end{array}$

Essai d'une ^{générale} des éléments
d'après les poids atomiques et
fonctions chimiques de L. Landolt
et les données universelles de L. Landolt

JS $\frac{\pi}{17}$ 69.

[illegible]

Arthur L. Lodge, & Vancouver. What expense by -

Lavoisier's 'Table of Simple substances' (1789); de Chancourtois' 'Telluric Screw' (1862); Mendeleev's hand-written periodic table (1869)



OPCW

1871

[illegible]

	H	
Li Be B C N O F	Na Mg Al Si P S Cl	
K Ca - Ti V Cr Mn Fe Co Ni Cu Zn - As Se Br		
Rb Sr Y Zr Nb Mo - Ru Rh Pd Ag Cd In Sn Sb Te I		
Cs Ba Dy Ce		
- Er La Ta W - Os Ir Pt Au Hg Tl Pb Bi -		
- Th U		

Group 0	I a b	II a b	III a b	IV a b	V a b	VI a b	VII a b	VIII
	H 1							
He 2	Li 3	Be 4	B 5	C 6	N 7	O 8	F 9	
Ne 10	Na 11	Mg 12	Al 13	Si 14	P 15	S 16	Cl 17	
Ar 18	K 19 Cu 29	Ca 20 Zn 30	Sc 21 Ga 31	Ti 22 Ge 32	V 23 As 33	Cr 24 Se 34	Mn 25 Br 35	Fe 26, Co 27, Ni 28
Kr 36	Rb 37 Ag 47	Sr 38 Cd 48	Y 39 In 49	Zr 40 Sn 50	Nb 41 Sb 51	Mo 42 Te 52	– I 53	Ru 44, Rh 45, Pd 46
Xe 54	Cs 55 Au 79	Ba 56 Hg 80	57-71* Tl 81	Hf 72 Pb 82	Ta 73 Bi 83	W 74 Po 84	Re 75 –	Os 76, Ir 77, Pt 78
Rn 86	–	Ra 88	Ac 89	Th 90	Pa 91	U 92		

1914

<div><div><div><div><div>1</div><div>H</div><div>Hydrogen</div><div>1.00794 (1)</div></div><div><div>3</div><div>Li</div><div>Lithium</div><div>6.941</div></div><div><div>11</div><div>Na</div><div>Sodium</div><div>22.98976928</div></div></div><div><div>2</div><div>He</div><div>Helium</div><div>4.002602</div></div><div><div>4</div><div>Be</div><div>Beryllium</div><div>9.012182</div></div><div><div>12</div><div>Mg</div><div>Magnesium</div><div>24.304</div></div></div><div><div>5</div><div>B</div><div>Boron</div><div>10.811</div></div><div><div>6</div><div>C</div><div>Carbon</div><div>12.0107</div></div><div><div>7</div><div>N</div><div>Nitrogen</div><div>14.00643</div></div><div><div>8</div><div>O</div><div>Oxygen</div><div>15.999</div></div><div><div>9</div><div>F</div><div>Fluorine</div><div>18.9984032</div></div><div><div>10</div><div>Ne</div><div>Neon</div><div>20.1797</div></div></div> <div><div>13</div><div>Al</div><div>Aluminum</div><div>26.9815385</div></div> <div><div>14</div><div>Si</div><div>Silicon</div><div>28.08558</div></div> <div><div>15</div><div>P</div><div>Phosphorus</div><div>30.973762</div></div> <div><div>16</div><div>S</div><div>Sulfur</div><div>32.06</div></div> <div><div>17</div><div>Cl</div><div>Chlorine</div><div>35.45</div></div> <div><div>18</div><div>Ar</div><div>Argon</div><div>39.948</div></div> <tr><td colspan="18"><div><div><div><div><div>19</div><div>K</div><div>Potassium</div><div>39.0983</div></div><div><div>27</div><div>Rb</div><div>Rubidium</div><div>85.4678</div></div><div><div>37</div><div>Fr</div><div>Francium</div><div>223</div></div></div><div><div>20</div><div>Ca</div><div>Calcium</div><div>40.078</div></div><div><div>28</div><div>Sr</div><div>Strontium</div><div>87.62</div></div><div><div>38</div><div>Ra</div><div>Radium</div><div>226</div></div></div><div><div>21</div><div>Sc</div><div>Scandium</div><div>44.955912</div></div><div><div>29</div><div>Y</div><div>Yttrium</div><div>88.90584</div></div><div><div>39</div><div>Ac</div><div>Actinium</div><div>227</div></div></div><div><div>22</div><div>Ti</div><div>Titanium</div><div>47.88</div></div><div><div>30</div><div>Zr</div><div>Zirconium</div><div>91.224</div></div><div><div>40</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div><div><div>41</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div><div><div>42</div><div>Tc</div><div>Technetium</div><div>98</div></div><div><div>43</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div><div><div>44</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div><div><div>45</div><div>Pd</div><div>Palladium</div><div>106.42</div></div><div><div>46</div><div>Ag</div><div>Silver</div><div>107.8682</div></div><div><div>47</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div><div><div>48</div><div>In</div><div>Indium</div><div>114.818</div></div><div><div>49</div><div>Sn</div><div>Tin</div><div>118.710</div></div><div><div>50</div><div>Sb</div><div>Antimony</div><div>121.757</div></div><div><div>51</div><div>Te</div><div>Tellurium</div><div>127.6</div></div><div><div>52</div><div>I</div><div>Iodine</div><div>126.905</div></div><div><div>53</div><div>Xe</div><div>Xenon</div><div>131.29</div></div><div><div>23</div><div>V</div><div>Vanadium</div><div>50.9415</div></div><div><div>31</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div><div><div>41</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div><div><div>43</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div><div><div>45</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div><div><div>46</div><div>Pd</div><div>Palladium</div><div>106.42</div></div><div><div>47</div><div>Ag</div><div>Silver</div><div>107.8682</div></div><div><div>48</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div><div><div>49</div><div>In</div><div>Indium</div><div>114.818</div></div><div><div>50</div><div>Sn</div><div>Tin</div><div>118.710</div></div><div><div>51</div><div>Sb</div><div>Antimony</div><div>121.757</div></div><div><div>52</div><div>Te</div><div>Tellurium</div><div>127.6</div></div><div><div>53</div><div>I</div><div>Iodine</div><div>126.905</div></div><div><div>54</div><div>Xe</div><div>Xenon</div><div>131.29</div></div><div><div>24</div><div>Cr</div><div>Chromium</div><div>51.9961</div></div><div><div>32</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div><div><div>42</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div><div><div>44</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div><div><div>46</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div><div><div>47</div><div>Pd</div><div>Palladium</div><div>106.42</div></div><div><div>48</div><div>Ag</div><div>Silver</div><div>107.8682</div></div><div><div>49</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div><div><div>50</div><div>In</div><div>Indium</div><div>114.818</div></div><div><div>51</div><div>Sb</div><div>Antimony</div><div>121.757</div></div><div><div>52</div><div>Te</div><div>Tellurium</div><div>127.6</div></div><div><div>53</div><div>I</div><div>Iodine</div><div>126.905</div></div><div><div>54</div><div>Xe</div><div>Xenon</div><div>131.29</div></div><div><div>25</div><div>Mn</div><div>Manganese</div><div>54.938045</div></div><div><div>33</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div><div><div>43</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div><div><div>45</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div><div><div>47</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div><div><div>48</div><div>Pd</div><div>Palladium</div><div>106.42</div></div><div><div>49</div><div>Ag</div><div>Silver</div><div>107.8682</div></div><div><div>50</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div><div><div>51</div><div>In</div><div>Indium</div><div>114.818</div></div><div><div>52</div><div>Sb</div><div>Antimony</div><div>121.757</div></div><div><div>53</div><div>Te</div><div>Tellurium</div><div>127.6</div></div><div><div>54</div><div>Xe</div><div>Xenon</div><div>131.29</div></div><div><div>26</div><div>Fe</div><div>Iron</div><div>55.845</div></div><div><div>34</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div><div><div>44</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div><div><div>46</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div><div><div>48</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div><div><div>49</div><div>Pd</div><div>Palladium</div><div>106.42</div></div><div><div>50</div><div>Ag</div><div>Silver</div><div>107.8682</div></div><div><div>51</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div><div><div>52</div><div>In</div><div>Indium</div><div>114.818</div></div><div><div>53</div><div>Sb</div><div>Antimony</div><div>121.757</div></div><div><div>54</div><div>Xe</div><div>Xenon</div><div>131.29</div></div><div><div>27</div><div>Co</div><div>Cobalt</div><div>58.933195</div></div><div><div>35</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div><div><div>45</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div><div><div>47</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div><div><div>49</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div><div><div>50</div><div>Pd</div><div>Palladium</div><div>106.42</div></div><div><div>51</div><div>Ag</div><div>Silver</div><div>107.8682</div></div><div><div>52</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div><div><div>53</div><div>In</div><div>Indium</div><div>114.818</div></div><div><div>54</div><div>Sb</div><div>Antimony</div><div>121.757</div></div><div><div>55</div><div>Te</div><div>Tellurium</div><div>127.6</div></div><div><div>56</div><div>I</div><div>Iodine</div><div>126.905</div></div><div><div>57</div><div>Xe</div><div>Xenon</div><div>131.29</div></div><div><div>28</div><div>Ni</div><div>Nickel</div><div>58.6934</div></div><div><div>36</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div><div><div>46</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div><div><div>48</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div><div><div>50</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div><div><div>51</div><div>Pd</div><div>Palladium</div><div>106.42</div></div><div><div>52</div><div>Ag</div><div>Silver</div><div>107.8682</div></div><div><div>53</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div><div><div>54</div><div>In</div><div>Indium</div><div>114.818</div></div><div><div>55</div><div>Sb</div><div>Antimony</div><div>121.757</div></div><div><div>56</div><div>Te</div><div>Tellurium</div><div>127.6</div></div><div><div>57</div><div>I</div><div>Iodine</div><div>126.905</div></div><div><div>58</div><div>Xe</div><div>Xenon</div><div>131.29</div></div><div><div>29</div><div>Cu</div><div>Copper</div><div>63.546</div></div><div><div>37</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div><div><div>47</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div><div><div>49</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div><div><div>51</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div><div><div>52</div><div>Pd</div><div>Palladium</div><div>106.42</div></div><div><div>53</div><div>Ag</div><div>Silver</div><div>107.8682</div></div><div><div>54</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div><div><div>55</div><div>In</div><div>Indium</div><div>114.818</div></div><div><div>56</div><div>Sb</div><div>Antimony</div><div>121.757</div></div><div><div>57</div><div>Te</div><div>Tellurium</div><div>127.6</div></div><div><div>58</div><div>I</div><div>Iodine</div><div>126.905</div></div><div><div>59</div><div>Xe</div><div>Xenon</div><div>131.29</div></div><div><div>30</div><div>Zn</div><div>Zinc</div><div>65.38</div></div><div><div>38</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div><div><div>48</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div><div><div>50</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div><div><div>52</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div><div><div>53</div><div>Pd</div><div>Palladium</div><div>106.42</div></div><div><div>54</div><div>Ag</div><div>Silver</div><div>107.8682</div></div><div><div>55</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div><div><div>56</div><div>In</div><div>Indium</div><div>114.818</div></div><div><div>57</div><div>Sb</div><div>Antimony</div><div>121.757</div></div><div><div>58</div><div>Te</div><div>Tellurium</div><div>127.6</div></div><div><div>59</div><div>I</div><div>Iodine</div><div>126.905</div></div><div><div>60</div><div>Xe</div><div>Xenon</div><div>131.29</div></div><div><div>31</div><div>Ga</div><div>Gallium</div><div>70.620</div></div><div><div>39</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div><div><div>49</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div><div><div>51</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div><div><div>53</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div><div><div>54</div><div>Pd</div><div>Palladium</div><div>106.42</div></div><div><div>55</div><div>Ag</div><div>Silver</div><div>107.8682</div></div><div><div>56</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div><div><div>57</div><div>In</div><div>Indium</div><div>114.818</div></div><div><div>58</div><div>Sb</div><div>Antimony</div><div>121.757</div></div><div><div>59</div><div>Te</div><div>Tellurium</div><div>127.6</div></div><div><div>60</div><div>I</div><div>Iodine</div><div>126.905</div></div><div><div>61</div><div>Xe</div><div>Xenon</div><div>131.29</div></div><div><div>32</div><div>Ge</div><div>Germanium</div><div>72.630</div></div><div><div>40</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div><div><div>50</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div><div><div>52</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div><div><div>54</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div><div><div>55</div><div>Pd</div><div>Palladium</div><div>106.42</div></div><div><div>56</div><div>Ag</div><div>Silver</div><div>107.8682</div></div><div><div>57</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div><div><div>58</div><div>In</div><div>Indium</div><div>114.818</div></div><div><div>59</div><div>Sb</div><div>Antimony</div><div>121.757</div></div><div><div>60</div><div>Te</div><div>Tellurium</div><div>127.6</div></div><div><div>61</div><div>I</div><div>Iodine</div><div>126.905</div></div><div><div>62</div><div>Xe</div><div>Xenon</div><div>131.29</div></div><div><div>33</div><div>As</div><div>Arsenic</div><div>74.9216</div></div><div><div>41</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div><div><div>51</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div><div><div>53</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div><div><div>55</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div><div><div>56</div><div>Pd</div><div>Palladium</div><div>106.42</div></div><div><div>57</div><div>Ag</div><div>Silver</div><div>107.8682</div></div><div><div>58</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div><div><div>59</div><div>In</div><div>Indium</div><div>114.818</div></div><div><div>60</div><div>Sb</div><div>Antimony</div><div>121.757</div></div><div><div>61</div><div>Te</div><div>Tellurium</div><div>127.6</div></div><div><div>62</div><div>I</div><div>Iodine</div><div>126.905</div></div><div><div>63</div><div>Xe</div><div>Xenon</div><div>131.29</div></div><div><div>34</div><div>Se</div><div>Selenium</div><div>78.96</div></div><div><div>42</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div><div><div>52</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div><div><div>54</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div><div><div>56</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div><div><div>57</div><div>Pd</div><div>Palladium</div><div>106.42</div></div><div><div>58</div><div>Ag</div><div>Silver</div><div>107.8682</div></div><div><div>59</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div><div><div>60</div><div>In</div><div>Indium</div><div>114.818</div></div><div><div>61</div><div>Sb</div><div>Antimony</div><div>121.757</div></div><div><div>62</div><div>Te</div><div>Tellurium</div><div>127.6</div></div><div><div>63</div><div>I</div><div>Iodine</div><div>126.905</div></div><div><div>64</div><div>Xe</div><div>Xenon</div><div>131.29</div></div><div><div>35</div><div>Br</div><div>Bromine</div><div>79.904</div></div><div><div>43</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div><div><div>53</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div><div><div>55</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div><div><div>57</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div><div><div>58</div><div>Pd</div><div>Palladium</div><div>106.42</div></div><div><div>59</div><div>Ag</div><div>Silver</div><div>107.8682</div></div><div><div>60</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div><div><div>61</div><div>In</div><div>Indium</div><div>114.818</div></div><div><div>62</div><div>Sb</div><div>Antimony</div><div>121.757</div></div><div><div>63</div><div>Te</div><div>Tellurium</div><div>127.6</div></div><div><div>64</div><div>I</div><div>Iodine</div><div>126.905</div></div><div><div>65</div><div>Xe</div><div>Xenon</div><div>131.29</div></div><div><div>36</div><div>Kr</div><div>Krypton</div><div>83.798</div></div><div><div>44</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div><div><div>54</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div><div><div>56</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div><div><div>58</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div><div><div>59</div><div>Pd</div><div>Palladium</div><div>106.42</div></div><div><div>60</div><div>Ag</div><div>Silver</div><div>107.8682</div></div><div><div>61</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div><div><div>62</div><div>In</div><div>Indium</div><div>114.818</div></div><div><div>63</div><div>Sb</div><div>Antimony</div><div>121.757</div></div><div><div>64</div><div>Te</div><div>Tellurium</div><div>127.6</div></div><div><div>65</div><div>I</div><div>Iodine</div><div>126.905</div></div><div><div>66</div><div>Xe</div><div>Xenon</div><div>131.29</div></div><div><div>38</div><div>Br</div><div>Bromine</div><div>79.904</div></div><div><div>46</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div><div><div>56</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div><div><div>58</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div><div><div>60</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div><div><div>61</div><div>Pd</div><div>Palladium</div><div>106.42</div></div><div><div>62</div><div>Ag</div><div>Silver</div><div>107.8682</div></div><div><div>63</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div><div><div>64</div><div>In</div><div>Indium</div><div>114.818</div></div><div><div>65</div><div>Sb</div><div>Antimony</div><div>121.757</div></div><div><div>66</div><div>Te</div><div>Tellurium</div><div>127.6</div></div><div><div>67</div><div>I</div><div>Iodine</div><div>126.905</div></div><div><div>68</div><div>Xe</div><div>Xenon</div><div>131.29</div></div><div><div>39</div><div>Kr</div><div>Krypton</div><div>83.798</div></div><div><div>47</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div><div><div>57</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div><div><div>59</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div><div><div>61</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div><div><div>62</div><div>Pd</div><div>Palladium</div><div>106.42</div></div><div><div>63</div><div>Ag</div><div>Silver</div><div>107.8682</div></div><div><div>64</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div><div><div>65</div><div>In</div><div>Indium</div><div>114.818</div></div><div><div>66</div><div>Sb</div><div>Antimony</div><div>121.757</div></div><div><div>67</div><div>Te</div><div>Tellurium</div><div>127.6</div></div><div><div>68</div><div>I</div><div>Iodine</div><div>126.905</div></div><div><div>69</div><div>Xe</div><div>Xenon</div><div>131.29</div></div><div><div>40</div><div>Kr</div><div>Krypton</div><div>83.798</div></div><div><div>48</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div><div><div>58</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div><div><div>60</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div><div><div>62</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div><div><div>63</div><div>Pd</div><div>Palladium</div><div>106.42</div></div><div><div>64</div><div>Ag</div><div>Silver</div><div>107.8682</div></div><div><div>65</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div><div><div>66</div><div>In</div><div>Indium</div><div>114.818</div></div><div><div>67</div><div>Sb</div><div>Antimony</div><div>121.757</div></div><div><div>68</div><div>Te</div><div>Tellurium</div><div>127.6</div></div><div><div>69</div><div>I</div><div>Iodine</div><div>126.905</div></div><div><div>70</div><div>Xe</div><div>Xenon</div><div>131.29</div></div><div><div>41</div><div>Kr</div><div>Krypton</div><div>83.798</div></div><div><div>49</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div><div><div>59</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div><div><div>61</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div><div><div>63</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div><div><div>64</div><div>Pd</div><div>Palladium</div><div>106.42</div></div><div><div>65</div><div>Ag</div><div>Silver</div><div>107.8682</div></div><div><div>66</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div><div><div>67</div><div>In</div><div>Indium</div><div>114.818</div></div><div><div>68</div><div>Sb</div><div>Antimony</div><div>121.757</div></div><div><div>69</div><div>Te</div><div>Tellurium</div><div>127.6</div></div><div><div>70</div><div>I</div><div>Iodine</div><div>126.905</div></div><div><div>71</div><div>Xe</div><div>Xenon</div><div>131.29</div></div><div><div>42</div><div>Kr</div><div>Krypton</div><div>83.798</div></div><div><div>50</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div><div><div>60</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div><div><div>62</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div><div><div>64</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div><div><div>65</div><div>Pd</div><div>Palladium</div><div>106.42</div></div><div><div>66</div><div>Ag</div><div>Silver</div><div>107.8682</div></div><div><div>67</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div><div><div>68</div><div>In</div><div>Indium</div><div>114.818</div></div><div><div>69</div><div>Sb</div><div>Antimony</div><div>121.757</div></div><div><div>70</div><div>Te</div><div>Tellurium</div><div>127.6</div></div><div><div>71</div><div>I</div><div>Iodine</div><div>126.905</div></div><div><div>72</div><div>Xe</div><div>Xenon</div><div>131.29</div></div><div><div>43</div><div>Kr</div><div>Krypton</div><div>83.798</div></div><div><div>51</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div><div><div>61</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div><div><div>63</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div><div><div>65</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div><div><div>66</div><div>Pd</div><div>Palladium</div><div>106.42</div></div><div><div>67</div><div>Ag</div><div>Silver</div><div>107.8682</div></div><div><div>68</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div><div><div>69</div><div>In</div><div>Indium</div><div>114.818</div></div><div><div>70</div><div>Sb</div><div>Antimony</div><div>121.757</div></div><div><div>71</div><div>Te</div><div>Tellurium</div><div>127.6</div></div><div><div>72</div><div>I</div><div>Iodine</div><div>126.905</div></div><div><div>73</div><div>Xe</div><div>Xenon</div><div>131.29</div></div><div><div>44</div><div>Kr</div><div>Krypton</div><div>83.798</div></div><div><div>52</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div><div><div>62</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div><div><div>64</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div><div><div>66</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div><div><div>67</div><div>Pd</div><div>Palladium</div><div>106.42</div></div><div><div>68</div><div>Ag</div><div>Silver</div><div>107.8682</div></div><div><div>69</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div><div><div>70</div><div>In</div><div>Indium</div><div>114.818</div></div><div><div>71</div><div>Sb</div><div>Antimony</div><div>121.757</div></div><div><div>72</div><div>Te</div><div>Tellurium</div><div>127.6</div></div><div><div>73</div><div>I</div><div>Iodine</div><div>126.905</div></div><div><div>74</div><div>Xe</div><div>Xenon</div><div>131.29</div></div><div><div>45</div><div>Kr</div><div>Krypton</div><div>83.798</div></div><div><div>53</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div><div><div>63</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div><div><div>65</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div><div><div>67</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div><div><div>68</div><div>Pd</div><div>Palladium</div><div>106.42</div></div><div><div>69</div><div>Ag</div><div>Silver</div><div>107.8682</div></div><div><div>70</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div><div><div>71</div><div>In</div><div>Indium</div><div>114.818</div></div><div><div>72</div><div>Sb</div><div>Antimony</div><div>121.757</div></div><div><div>73</div><div>Te</div><div>Tellurium</div><div>127.6</div></div><div><div>74</div><div>I</div><div>Iodine</div><div>126.905</div></div><div><div>75</div><div>Xe</div><div>Xenon</div><div>131.29</div></div><div><div>46</div><div>Kr</div><div>Krypton</div><div>83.798</div></div><div><div>54</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div><div><div>64</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div><div><div>66</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div><div><div>68</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div><div><div>69</div><div>Pd</div><div>Palladium</div><div>106.42</div></div><div><div>70</div><div>Ag</div><div>Silver</div><div>107.8682</div></div><div><div>71</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div><div><div>72</div><div>In</div><div>Indium</div><div>114.818</div></div><div><div>73</div><div>Sb</div><div>Antimony</div><div>121.757</div></div><div><div>74</div><div>Te</div><div>Tellurium</div><div>127.6</div></div><div><div>75</div><div>I</div><div>Iodine</div><div>126.905</div></div><div><div>76</div><div>Xe</div><div>Xenon</div><div>131.29</div></div><div><div>47</div><div>Kr</div><div>Krypton</div><div>83.798</div></div><div><div>55</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div><div><div>65</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div><div><div>67</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div><div><div>69</</div></div></td></tr>																		<div><div><div><div><div>19</div><div>K</div><div>Potassium</div><div>39.0983</div></div><div><div>27</div><div>Rb</div><div>Rubidium</div><div>85.4678</div></div><div><div>37</div><div>Fr</div><div>Francium</div><div>223</div></div></div><div><div>20</div><div>Ca</div><div>Calcium</div><div>40.078</div></div><div><div>28</div><div>Sr</div><div>Strontium</div><div>87.62</div></div><div><div>38</div><div>Ra</div><div>Radium</div><div>226</div></div></div><div><div>21</div><div>Sc</div><div>Scandium</div><div>44.955912</div></div><div><div>29</div><div>Y</div><div>Yttrium</div><div>88.90584</div></div><div><div>39</div><div>Ac</div><div>Actinium</div><div>227</div></div></div> <div><div>22</div><div>Ti</div><div>Titanium</div><div>47.88</div></div> <div><div>30</div><div>Zr</div><div>Zirconium</div><div>91.224</div></div> <div><div>40</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div> <div><div>41</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div> <div><div>42</div><div>Tc</div><div>Technetium</div><div>98</div></div> <div><div>43</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div> <div><div>44</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div> <div><div>45</div><div>Pd</div><div>Palladium</div><div>106.42</div></div> <div><div>46</div><div>Ag</div><div>Silver</div><div>107.8682</div></div> <div><div>47</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div> <div><div>48</div><div>In</div><div>Indium</div><div>114.818</div></div> <div><div>49</div><div>Sn</div><div>Tin</div><div>118.710</div></div> <div><div>50</div><div>Sb</div><div>Antimony</div><div>121.757</div></div> <div><div>51</div><div>Te</div><div>Tellurium</div><div>127.6</div></div> <div><div>52</div><div>I</div><div>Iodine</div><div>126.905</div></div> <div><div>53</div><div>Xe</div><div>Xenon</div><div>131.29</div></div> <div><div>23</div><div>V</div><div>Vanadium</div><div>50.9415</div></div> <div><div>31</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div> <div><div>41</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div> <div><div>43</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div> <div><div>45</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div> <div><div>46</div><div>Pd</div><div>Palladium</div><div>106.42</div></div> <div><div>47</div><div>Ag</div><div>Silver</div><div>107.8682</div></div> <div><div>48</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div> <div><div>49</div><div>In</div><div>Indium</div><div>114.818</div></div> <div><div>50</div><div>Sn</div><div>Tin</div><div>118.710</div></div> <div><div>51</div><div>Sb</div><div>Antimony</div><div>121.757</div></div> <div><div>52</div><div>Te</div><div>Tellurium</div><div>127.6</div></div> <div><div>53</div><div>I</div><div>Iodine</div><div>126.905</div></div> <div><div>54</div><div>Xe</div><div>Xenon</div><div>131.29</div></div> <div><div>24</div><div>Cr</div><div>Chromium</div><div>51.9961</div></div> <div><div>32</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div> <div><div>42</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div> <div><div>44</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div> <div><div>46</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div> <div><div>47</div><div>Pd</div><div>Palladium</div><div>106.42</div></div> <div><div>48</div><div>Ag</div><div>Silver</div><div>107.8682</div></div> <div><div>49</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div> <div><div>50</div><div>In</div><div>Indium</div><div>114.818</div></div> <div><div>51</div><div>Sb</div><div>Antimony</div><div>121.757</div></div> <div><div>52</div><div>Te</div><div>Tellurium</div><div>127.6</div></div> <div><div>53</div><div>I</div><div>Iodine</div><div>126.905</div></div> <div><div>54</div><div>Xe</div><div>Xenon</div><div>131.29</div></div> <div><div>25</div><div>Mn</div><div>Manganese</div><div>54.938045</div></div> <div><div>33</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div> <div><div>43</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div> <div><div>45</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div> <div><div>47</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div> <div><div>48</div><div>Pd</div><div>Palladium</div><div>106.42</div></div> <div><div>49</div><div>Ag</div><div>Silver</div><div>107.8682</div></div> <div><div>50</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div> <div><div>51</div><div>In</div><div>Indium</div><div>114.818</div></div> <div><div>52</div><div>Sb</div><div>Antimony</div><div>121.757</div></div> <div><div>53</div><div>Te</div><div>Tellurium</div><div>127.6</div></div> <div><div>54</div><div>Xe</div><div>Xenon</div><div>131.29</div></div> <div><div>26</div><div>Fe</div><div>Iron</div><div>55.845</div></div> <div><div>34</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div> <div><div>44</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div> <div><div>46</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div> <div><div>48</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div> <div><div>49</div><div>Pd</div><div>Palladium</div><div>106.42</div></div> <div><div>50</div><div>Ag</div><div>Silver</div><div>107.8682</div></div> <div><div>51</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div> <div><div>52</div><div>In</div><div>Indium</div><div>114.818</div></div> <div><div>53</div><div>Sb</div><div>Antimony</div><div>121.757</div></div> <div><div>54</div><div>Xe</div><div>Xenon</div><div>131.29</div></div> <div><div>27</div><div>Co</div><div>Cobalt</div><div>58.933195</div></div> <div><div>35</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div> <div><div>45</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div> <div><div>47</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div> <div><div>49</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div> <div><div>50</div><div>Pd</div><div>Palladium</div><div>106.42</div></div> <div><div>51</div><div>Ag</div><div>Silver</div><div>107.8682</div></div> <div><div>52</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div> <div><div>53</div><div>In</div><div>Indium</div><div>114.818</div></div> <div><div>54</div><div>Sb</div><div>Antimony</div><div>121.757</div></div> <div><div>55</div><div>Te</div><div>Tellurium</div><div>127.6</div></div> <div><div>56</div><div>I</div><div>Iodine</div><div>126.905</div></div> <div><div>57</div><div>Xe</div><div>Xenon</div><div>131.29</div></div> <div><div>28</div><div>Ni</div><div>Nickel</div><div>58.6934</div></div> <div><div>36</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div> <div><div>46</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div> <div><div>48</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div> <div><div>50</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div> <div><div>51</div><div>Pd</div><div>Palladium</div><div>106.42</div></div> <div><div>52</div><div>Ag</div><div>Silver</div><div>107.8682</div></div> <div><div>53</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div> <div><div>54</div><div>In</div><div>Indium</div><div>114.818</div></div> <div><div>55</div><div>Sb</div><div>Antimony</div><div>121.757</div></div> <div><div>56</div><div>Te</div><div>Tellurium</div><div>127.6</div></div> <div><div>57</div><div>I</div><div>Iodine</div><div>126.905</div></div> <div><div>58</div><div>Xe</div><div>Xenon</div><div>131.29</div></div> <div><div>29</div><div>Cu</div><div>Copper</div><div>63.546</div></div> <div><div>37</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div> <div><div>47</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div> <div><div>49</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div> <div><div>51</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div> <div><div>52</div><div>Pd</div><div>Palladium</div><div>106.42</div></div> <div><div>53</div><div>Ag</div><div>Silver</div><div>107.8682</div></div> <div><div>54</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div> <div><div>55</div><div>In</div><div>Indium</div><div>114.818</div></div> <div><div>56</div><div>Sb</div><div>Antimony</div><div>121.757</div></div> <div><div>57</div><div>Te</div><div>Tellurium</div><div>127.6</div></div> <div><div>58</div><div>I</div><div>Iodine</div><div>126.905</div></div> <div><div>59</div><div>Xe</div><div>Xenon</div><div>131.29</div></div> <div><div>30</div><div>Zn</div><div>Zinc</div><div>65.38</div></div> <div><div>38</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div> <div><div>48</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div> <div><div>50</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div> <div><div>52</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div> <div><div>53</div><div>Pd</div><div>Palladium</div><div>106.42</div></div> <div><div>54</div><div>Ag</div><div>Silver</div><div>107.8682</div></div> <div><div>55</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div> <div><div>56</div><div>In</div><div>Indium</div><div>114.818</div></div> <div><div>57</div><div>Sb</div><div>Antimony</div><div>121.757</div></div> <div><div>58</div><div>Te</div><div>Tellurium</div><div>127.6</div></div> <div><div>59</div><div>I</div><div>Iodine</div><div>126.905</div></div> 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<div><div>61</div><div>Xe</div><div>Xenon</div><div>131.29</div></div> <div><div>32</div><div>Ge</div><div>Germanium</div><div>72.630</div></div> <div><div>40</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div> <div><div>50</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div> <div><div>52</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div> <div><div>54</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div> <div><div>55</div><div>Pd</div><div>Palladium</div><div>106.42</div></div> <div><div>56</div><div>Ag</div><div>Silver</div><div>107.8682</div></div> <div><div>57</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div> <div><div>58</div><div>In</div><div>Indium</div><div>114.818</div></div> <div><div>59</div><div>Sb</div><div>Antimony</div><div>121.757</div></div> <div><div>60</div><div>Te</div><div>Tellurium</div><div>127.6</div></div> <div><div>61</div><div>I</div><div>Iodine</div><div>126.905</div></div> 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<div><div>65</div><div>Xe</div><div>Xenon</div><div>131.29</div></div> <div><div>36</div><div>Kr</div><div>Krypton</div><div>83.798</div></div> <div><div>44</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div> <div><div>54</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div> <div><div>56</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div> <div><div>58</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div> <div><div>59</div><div>Pd</div><div>Palladium</div><div>106.42</div></div> <div><div>60</div><div>Ag</div><div>Silver</div><div>107.8682</div></div> <div><div>61</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div> <div><div>62</div><div>In</div><div>Indium</div><div>114.818</div></div> <div><div>63</div><div>Sb</div><div>Antimony</div><div>121.757</div></div> <div><div>64</div><div>Te</div><div>Tellurium</div><div>127.6</div></div> <div><div>65</div><div>I</div><div>Iodine</div><div>126.905</div></div> 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<div><div>68</div><div>Xe</div><div>Xenon</div><div>131.29</div></div> <div><div>39</div><div>Kr</div><div>Krypton</div><div>83.798</div></div> <div><div>47</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div> <div><div>57</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div> <div><div>59</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div> <div><div>61</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div> <div><div>62</div><div>Pd</div><div>Palladium</div><div>106.42</div></div> <div><div>63</div><div>Ag</div><div>Silver</div><div>107.8682</div></div> <div><div>64</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div> <div><div>65</div><div>In</div><div>Indium</div><div>114.818</div></div> <div><div>66</div><div>Sb</div><div>Antimony</div><div>121.757</div></div> <div><div>67</div><div>Te</div><div>Tellurium</div><div>127.6</div></div> <div><div>68</div><div>I</div><div>Iodine</div><div>126.905</div></div> <div><div>69</div><div>Xe</div><div>Xenon</div><div>131.29</div></div> <div><div>40</div><div>Kr</div><div>Krypton</div><div>83.798</div></div> <div><div>48</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div> <div><div>58</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div> <div><div>60</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div> <div><div>62</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div> <div><div>63</div><div>Pd</div><div>Palladium</div><div>106.42</div></div> <div><div>64</div><div>Ag</div><div>Silver</div><div>107.8682</div></div> <div><div>65</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div> <div><div>66</div><div>In</div><div>Indium</div><div>114.818</div></div> <div><div>67</div><div>Sb</div><div>Antimony</div><div>121.757</div></div> <div><div>68</div><div>Te</div><div>Tellurium</div><div>127.6</div></div> <div><div>69</div><div>I</div><div>Iodine</div><div>126.905</div></div> <div><div>70</div><div>Xe</div><div>Xenon</div><div>131.29</div></div> <div><div>41</div><div>Kr</div><div>Krypton</div><div>83.798</div></div> <div><div>49</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div> <div><div>59</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div> <div><div>61</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div> <div><div>63</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div> <div><div>64</div><div>Pd</div><div>Palladium</div><div>106.42</div></div> <div><div>65</div><div>Ag</div><div>Silver</div><div>107.8682</div></div> <div><div>66</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div> <div><div>67</div><div>In</div><div>Indium</div><div>114.818</div></div> <div><div>68</div><div>Sb</div><div>Antimony</div><div>121.757</div></div> <div><div>69</div><div>Te</div><div>Tellurium</div><div>127.6</div></div> <div><div>70</div><div>I</div><div>Iodine</div><div>126.905</div></div> 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<div><div>72</div><div>Xe</div><div>Xenon</div><div>131.29</div></div> <div><div>43</div><div>Kr</div><div>Krypton</div><div>83.798</div></div> <div><div>51</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div> <div><div>61</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div> <div><div>63</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div> <div><div>65</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div> <div><div>66</div><div>Pd</div><div>Palladium</div><div>106.42</div></div> <div><div>67</div><div>Ag</div><div>Silver</div><div>107.8682</div></div> <div><div>68</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div> <div><div>69</div><div>In</div><div>Indium</div><div>114.818</div></div> <div><div>70</div><div>Sb</div><div>Antimony</div><div>121.757</div></div> <div><div>71</div><div>Te</div><div>Tellurium</div><div>127.6</div></div> <div><div>72</div><div>I</div><div>Iodine</div><div>126.905</div></div> 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<div><div>76</div><div>Xe</div><div>Xenon</div><div>131.29</div></div> <div><div>47</div><div>Kr</div><div>Krypton</div><div>83.798</div></div> <div><div>55</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div> <div><div>65</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div> <div><div>67</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div> <div><div>69</</div></div>																	
<div><div><div><div><div>19</div><div>K</div><div>Potassium</div><div>39.0983</div></div><div><div>27</div><div>Rb</div><div>Rubidium</div><div>85.4678</div></div><div><div>37</div><div>Fr</div><div>Francium</div><div>223</div></div></div><div><div>20</div><div>Ca</div><div>Calcium</div><div>40.078</div></div><div><div>28</div><div>Sr</div><div>Strontium</div><div>87.62</div></div><div><div>38</div><div>Ra</div><div>Radium</div><div>226</div></div></div><div><div>21</div><div>Sc</div><div>Scandium</div><div>44.955912</div></div><div><div>29</div><div>Y</div><div>Yttrium</div><div>88.90584</div></div><div><div>39</div><div>Ac</div><div>Actinium</div><div>227</div></div></div> <div><div>22</div><div>Ti</div><div>Titanium</div><div>47.88</div></div> <div><div>30</div><div>Zr</div><div>Zirconium</div><div>91.224</div></div> <div><div>40</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div> <div><div>41</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div> <div><div>42</div><div>Tc</div><div>Technetium</div><div>98</div></div> <div><div>43</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div> <div><div>44</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div> <div><div>45</div><div>Pd</div><div>Palladium</div><div>106.42</div></div> <div><div>46</div><div>Ag</div><div>Silver</div><div>107.8682</div></div> <div><div>47</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div> <div><div>48</div><div>In</div><div>Indium</div><div>114.818</div></div> <div><div>49</div><div>Sn</div><div>Tin</div><div>118.710</div></div> <div><div>50</div><div>Sb</div><div>Antimony</div><div>121.757</div></div> <div><div>51</div><div>Te</div><div>Tellurium</div><div>127.6</div></div> <div><div>52</div><div>I</div><div>Iodine</div><div>126.905</div></div> <div><div>53</div><div>Xe</div><div>Xenon</div><div>131.29</div></div> <div><div>23</div><div>V</div><div>Vanadium</div><div>50.9415</div></div> 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<div><div>24</div><div>Cr</div><div>Chromium</div><div>51.9961</div></div> <div><div>32</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div> <div><div>42</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div> <div><div>44</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div> <div><div>46</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div> <div><div>47</div><div>Pd</div><div>Palladium</div><div>106.42</div></div> <div><div>48</div><div>Ag</div><div>Silver</div><div>107.8682</div></div> <div><div>49</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div> <div><div>50</div><div>In</div><div>Indium</div><div>114.818</div></div> <div><div>51</div><div>Sb</div><div>Antimony</div><div>121.757</div></div> <div><div>52</div><div>Te</div><div>Tellurium</div><div>127.6</div></div> <div><div>53</div><div>I</div><div>Iodine</div><div>126.905</div></div> <div><div>54</div><div>Xe</div><div>Xenon</div><div>131.29</div></div> <div><div>25</div><div>Mn</div><div>Manganese</div><div>54.938045</div></div> <div><div>33</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div> <div><div>43</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div> <div><div>45</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div> <div><div>47</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div> <div><div>48</div><div>Pd</div><div>Palladium</div><div>106.42</div></div> <div><div>49</div><div>Ag</div><div>Silver</div><div>107.8682</div></div> <div><div>50</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div> <div><div>51</div><div>In</div><div>Indium</div><div>114.818</div></div> <div><div>52</div><div>Sb</div><div>Antimony</div><div>121.757</div></div> <div><div>53</div><div>Te</div><div>Tellurium</div><div>127.6</div></div> <div><div>54</div><div>Xe</div><div>Xenon</div><div>131.29</div></div> <div><div>26</div><div>Fe</div><div>Iron</div><div>55.845</div></div> <div><div>34</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div> <div><div>44</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div> <div><div>46</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div> <div><div>48</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div> <div><div>49</div><div>Pd</div><div>Palladium</div><div>106.42</div></div> <div><div>50</div><div>Ag</div><div>Silver</div><div>107.8682</div></div> <div><div>51</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div> <div><div>52</div><div>In</div><div>Indium</div><div>114.818</div></div> <div><div>53</div><div>Sb</div><div>Antimony</div><div>121.757</div></div> <div><div>54</div><div>Xe</div><div>Xenon</div><div>131.29</div></div> <div><div>27</div><div>Co</div><div>Cobalt</div><div>58.933195</div></div> <div><div>35</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div> <div><div>45</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div> <div><div>47</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div> <div><div>49</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div> <div><div>50</div><div>Pd</div><div>Palladium</div><div>106.42</div></div> <div><div>51</div><div>Ag</div><div>Silver</div><div>107.8682</div></div> <div><div>52</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div> <div><div>53</div><div>In</div><div>Indium</div><div>114.818</div></div> <div><div>54</div><div>Sb</div><div>Antimony</div><div>121.757</div></div> <div><div>55</div><div>Te</div><div>Tellurium</div><div>127.6</div></div> <div><div>56</div><div>I</div><div>Iodine</div><div>126.905</div></div> <div><div>57</div><div>Xe</div><div>Xenon</div><div>131.29</div></div> <div><div>28</div><div>Ni</div><div>Nickel</div><div>58.6934</div></div> <div><div>36</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div> <div><div>46</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div> <div><div>48</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div> <div><div>50</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div> <div><div>51</div><div>Pd</div><div>Palladium</div><div>106.42</div></div> <div><div>52</div><div>Ag</div><div>Silver</div><div>107.8682</div></div> <div><div>53</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div> <div><div>54</div><div>In</div><div>Indium</div><div>114.818</div></div> <div><div>55</div><div>Sb</div><div>Antimony</div><div>121.757</div></div> <div><div>56</div><div>Te</div><div>Tellurium</div><div>127.6</div></div> <div><div>57</div><div>I</div><div>Iodine</div><div>126.905</div></div> <div><div>58</div><div>Xe</div><div>Xenon</div><div>131.29</div></div> <div><div>29</div><div>Cu</div><div>Copper</div><div>63.546</div></div> <div><div>37</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div> <div><div>47</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div> 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<div><div>50</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div> <div><div>52</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div> <div><div>53</div><div>Pd</div><div>Palladium</div><div>106.42</div></div> <div><div>54</div><div>Ag</div><div>Silver</div><div>107.8682</div></div> <div><div>55</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div> <div><div>56</div><div>In</div><div>Indium</div><div>114.818</div></div> <div><div>57</div><div>Sb</div><div>Antimony</div><div>121.757</div></div> <div><div>58</div><div>Te</div><div>Tellurium</div><div>127.6</div></div> <div><div>59</div><div>I</div><div>Iodine</div><div>126.905</div></div> <div><div>60</div><div>Xe</div><div>Xenon</div><div>131.29</div></div> <div><div>31</div><div>Ga</div><div>Gallium</div><div>70.620</div></div> <div><div>39</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div> <div><div>49</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div> <div><div>51</div><div>Ru</div><div>Ruthenium</div><div>101.07</div></div> <div><div>53</div><div>Rh</div><div>Rhodium</div><div>102.9055</div></div> <div><div>54</div><div>Pd</div><div>Palladium</div><div>106.42</div></div> <div><div>55</div><div>Ag</div><div>Silver</div><div>107.8682</div></div> <div><div>56</div><div>Cd</div><div>Cadmium</div><div>112.411</div></div> <div><div>57</div><div>In</div><div>Indium</div><div>114.818</div></div> <div><div>58</div><div>Sb</div><div>Antimony</div><div>121.757</div></div> <div><div>59</div><div>Te</div><div>Tellurium</div><div>127.6</div></div> <div><div>60</div><div>I</div><div>Iodine</div><div>126.905</div></div> <div><div>61</div><div>Xe</div><div>Xenon</div><div>131.29</div></div> <div><div>32</div><div>Ge</div><div>Germanium</div><div>72.630</div></div> <div><div>40</div><div>Nb</div><div>Niobium</div><div>92.90638</div></div> <div><div>50</div><div>Mo</div><div>Molybdenum</div><div>95.94</div></div> 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2019



OPCW

Periodic Table of States Parties to the Chemical Weapons Convention

In Honour of the International Year of the Periodic Table of Chemical Elements 2019

[illegible]

Periodic Table of States Parties

- Names of the Elements are changed to States Parties Names
- Colour – Regional groups
- Only 118 elements but 193 States Parties
 - Is Diplomacy further advanced than Chemistry?
- Order is the Entry in to Force order
 - Firstly looked at Entry into Force date (several the same)
 - Secondly looked at Signing date (several the same)
 - Thirdly looked at Date of Deposit (unique)

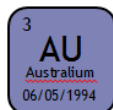


OPCW

What next?

Element Data Sheet?

3



Australium is the 3rd element on the Periodic Table of States Parties of the CWC.

AUSTRALIA

Signature	Date of Deposit, Ratification, Accession or Succession	Entry in Force
13 January 1993	6 May 1994	29 April 1997



Permanent Representative of Australia to the OPCW

H.E. Mr Matthew Neubeus
Ambassador Extraordinary and Plenipotentiary



Presentation of Credentials: 29 August 2016

Caracal 4
2517 KH Dg Hague
The Netherlands
<https://netherlands.em.basny.gov.au/thaq/home.html>

Department of Foreign Affairs and Trade – Australian Safeguards and Non-Proliferation Office (ASNO) – Chemical Weapons Convention Implementation Section

ASNO, R.G. Casey Building, John McEwen Crescent
Barton ACT 0221
Australia

<https://dfat.gov.au/international-relations/security/non-proliferation/disarmament-of-arms-control/chemical-weapons/cwc/Pages/australia-national-authority-for-the-chemical-weapons-convention.aspx>

Australia is the largest country in Oceania and Australia's capital is Canberra, and its largest city is Sydney. The country's other major metropolitan areas are Melbourne, Brisbane, Perth and Adelaide. Australia was inhabited by Indigenous Australians for about 60,000 years before the first British settlement in the late 18th century. Australia has a landmass of 7,617,950 square kilometres (2,941,300 sq mi). A megadiverse country, its size gives it a wide variety of landscapes, with deserts in the centre, tropical rainforests in the north-east and mountain ranges in the south-east. Australia is a highly developed country, with the world's 13th largest economy. Australia is a federal parliamentary constitutional monarchy. Australia is a member of the United Nations, G20, Commonwealth of Nations, ANZUS, Organisation for Economic Co-operation and Development (OECD), World Trade Organization, Asia-Pacific Economic Cooperation, Pacific Islands Forum and the ASEAN-BLIA-6 mechanism.



Interactive Table on Website?

Periodic Table of States Parties to the Chemical Weapons Convention
In Honour of the International Year of the Periodic Table of Chemical Elements 2019

LEGEND

- Western Europe and Other States (WEDG)
- Western Europe
- Africa
- Latin America and the Caribbean (IGRLAC)
- Asia

OPCW

IUPAC

INTERNATIONAL UNION OF PURE AND APPLIED CHEMISTRY

INTERNATIONAL UNION OF PHYSICAL AND APPLIED CHEMISTRY

INTERNATIONAL UNION OF THEORETICAL AND APPLIED CHEMISTRY

Who would like to join in?



OPCW

Links:

- www.iypt2019.org/
- iupac.org/
- www.opcw.org/
- www.opcw.org/resources/science-and-technology

- Download the Periodic Table of the States Parties

www.opcw.org/sites/default/files/documents/2018/11/Periodic%20Table%20of%20States%20Parties%20-%20Building.pdf (OPCW Building Background)

www.opcw.org/sites/default/files/documents/2018/11/Periodic%20Table%20of%20States%20Parties%20-%20flags.pdf (OPCW Flags Background)



OPCW



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منظمة حظر الأسلحة الكيميائية

禁止化学武器组织

Organisation for the Prohibition of Chemical Weapons

Organisation pour l'Interdiction des Armes Chimiques

Организация по запрещению химического оружия

Organización para la Prohibición de las Armas Químicas